

GRAYANOSIDE C, A NEW DITERPENE GLUCOSIDE FROM *LEUCOTHOE GRAYANA*

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Abstract—A new diterpene glucoside has been isolated from *Leucothoe grayana*. Its structure was elucidated by X-ray diffraction analysis of a dehydrated product of the aglycone and conversion of the same to leucothol A.

INTRODUCTION

In previous papers [1, 2] we reported the isolation of two grayanoid glucosides, grayanoside A and B, from *Leucothoe grayana*. This paper describes the isolation and structure determination of another new grayanoid glucoside, grayanoside C, from the same plant.

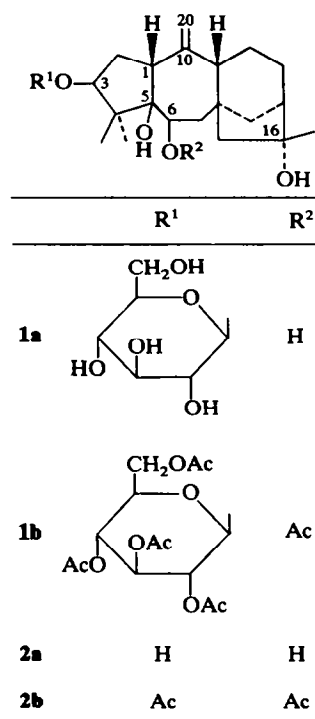
RESULTS AND DISCUSSION

Grayanoside C (**1a**), viscous syrup, was isolated from the *n*-BuOH-soluble fraction of an MeOH extract as well as grayanoside A and B. Acetylation of **1a** gave a pentaacetate (**1b**), $C_{36}H_{52}O_{14}$, mp 220–222°.

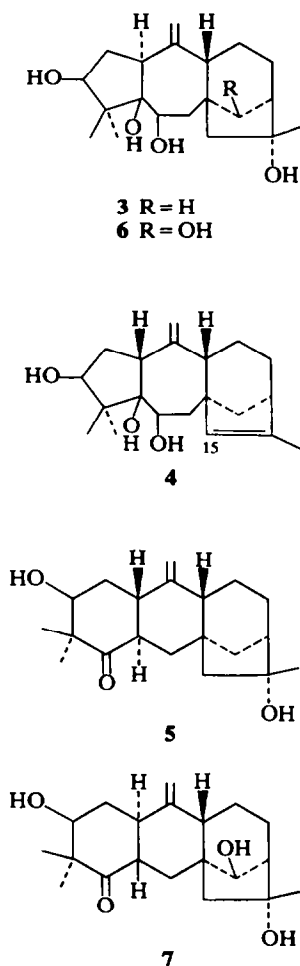
Acid hydrolysis of **1a** yielded glucose, but its aglycone could not be obtained. Enzymatic hydrolysis of **1a** with naringinase gave a genuine aglycone (**2a**), $C_{20}H_{30}O_4$, mp 232–233° (decomp.). Acetylation of **2a** yielded a diacetate (**2b**), syrup. These findings and comparison of the spectral data (see Experimental) suggested that **2a** has the same composition and functional groups as grayanotoxin-XVIII (**3**) [2]: three tertiary methyls, six methylenes, three methines, two quaternary carbons, two secondary hydroxyls, two tertiary hydroxyls and a vinylidene.

To confirm the structure, we tried to undertake an X-ray diffraction analysis of the aglycone (**2a**), but a suitable crystal could not be obtained. Then **2a** was refluxed with anhydrous $CuSO_4$ in dioxane to yield a monodehydrated product (**4**), $C_{20}H_{30}O_3$, mp 205–206° (decomp.), which was submitted for X-ray analysis. The stereoscopic drawings of the molecules, which revealed the chemical structure of **4**, 3 β ,5 β ,6 β -trihydroxy-(1 β H)-grayana-10(20),15-diene are shown in Fig. 1. Hence the structure of the original aglycone (**2a**) was assumed to be 3 β ,5 β ,6 β ,16 ξ -tetrahydroxy-(1 β H)-grayan-10(20)-ene. At this stage, however, the configuration of 16-hydroxyl group, and the glucosidation pattern and position in the glucoside (**1a**) remained uncertain.

The aglycone (**2a**) was treated with methanesulfonyl



chloride to give a monodehydrated product (**5**), $C_{20}H_{30}O_3$, mp 245°, other than **4**. The IR spectrum of **5** showed a carbonyl absorption (1702 cm^{-1}) and the 1H NMR revealed the presence of three tertiary methyls, one secondary hydroxyl group and one exomethylene group. Hikino and his coworkers reported that grayanotoxin-II (**6**) reacted with *p*-toluenesulfonyl chloride to yield a rearranged product (**7**), whose CD curve exhibited a positive Cotton effect [4]. Since the CD curve of **5** showed a negative Cotton effect and the starting material (**2a**) had 1 β -configuration, **5** must have a rearranged six-six-membered A–B ring and 1 β ,6 α -configuration [4].



From these findings, the structure of **5** was deduced to be leucothol A, which was previously isolated from *Leucothoe grayana* [5], and was identified. Consequently the 16-hydroxyl group of the aglycone (**2a**) should have an α -configuration like **5**.

In the ^1H NMR spectra the anomeric proton of **1a** and **1b** was observed at δ 4.88 (d , $J = 7$ Hz) and 4.58 (d , $J = 7$ Hz), respectively, so **1a** must be a β -D-glucoside. The glucosidation position was determined by the ^{13}C NMR spectra. The ^{13}C NMR signals were assigned by means of single-frequency off-resonance decoupling, selective proton decoupling and by comparing the spectra of several analogous compounds. In general, carbinyl carbon (α -carbon) signals of aglycone alcohols are displaced by 5.5–10 ppm on

glucosidation [6, 7]. The C-3 signal of the aglycone (**2a**) appeared at δ 83.6, while that of the glucoside (**1a**) was observed at δ 92.0. Other carbinyl carbon signals scarcely shifted on glucosidation. Hence β -D-glucose must be bound at C-3 of the aglycone (**2a**).

From the above data, grayanoside C (**1a**) was concluded to be 3-O-(β -D-glucopyranosyl)-(1 β H)-grayanotoxin-XVIII. This is the first grayanoid glucoside, epimerized at C-1, to be found in nature. In the course of our investigation, Furusaki and his co-workers reported the isolation of grayathol A from the same plant as ours, whose structure was identical with **4** [8].

EXPERIMENTAL

Mps were uncorr. ^1H NMR spectra were measured at 100 MHz. ^{13}C NMR spectra were measured at 15 MHz. The δ values are expressed in ppm downfield from TMS int. standard. MS (20 eV) were taken with a direct inlet.

Isolation of 1a. **1a** was obtained in the course of isolation of grayanoside A and B [1, 2]. Activated charcoal column chromatography of the *n*-BuOH extract afforded crude **1a** in the fractions eluted with MeOH–H₂O (8:2 to 9:1), 5 g of which was chromatographed on a Si gel column. The CHCl₃–MeOH (9:1) eluate gave **1a** (2.2 g), which was purified by silanised Si gel PLC (developing solvent: MeOH–H₂O (1:1)).

Grayanoside C (1a). Viscous syrup. $[\alpha]_D^{25} + 6.00^\circ$ (MeOH c 2.00). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3390, 1075. ^1H NMR ($\text{C}_5\text{D}_5\text{N}$): δ 1.17, 1.50, 1.70 (each 3H, s), 3.8–4.5 (many protons), 4.88 (1H, d , $J = 7$ Hz), 5.26 (s). ^{13}C NMR ($\text{C}_5\text{D}_5\text{N}$): δ 21.1 (Me), 23.4 (Me), 24.8 (Me), 26.7 (C-11 and C-12), 35.2 (C-2), 35.6 (C-14), 45.7 (C-8), 48.8 (C-7), 49.6 ($-\text{CH}-$), 51.5 (C-4), 55.3 ($-\text{CH}-$), 57.3 (C-15), 59.0 ($-\text{CH}-$), 62.7 (glucose C-6), 71.2 (C-6), 71.6 (glucose C-4), 75.6 (glucose C-2), 78.6 (C-16 and glucose C-3, C-5), 86.1 (C-5), 92.0 (C-3), 105.0 (glucose C-1), 110.3 ($-\text{C}=\text{CH}_2$), 153.9 ($-\text{C}=\text{CH}_2$).

Pentaacetylgrayanoside C (1b). Treatment of **1a** with Ac₂O–Py overnight at room temp. gave **1b**, mp 220–222° (MeOH). (Found: C, 60.55; H, 7.45. Calc. for $\text{C}_{36}\text{H}_{52}\text{O}_{14}$: C, 61.00; H, 7.39%). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3545, 1755, 1230. ^1H NMR (CDCl_3): δ 0.81, 1.07, 1.36 (each 3H, s), 2.03 (3H \times 3, s), 2.07 (3H \times 2, s), 4.58 (1H, d , $J = 7$ Hz).

Acid hydrolysis of 1a. A soln of **1a** (4 mg) in dioxane (1 ml) and 5% H₂SO₄ (2 ml) was heated for 2.5 hr on a steam bath. The mixture was cooled, diluted with H₂O (10 ml) and extracted with EtOAc. The EtOAc extract was evapd *in vacuo* to give a complex mixture. The aq. layer was treated with Amberlite CG-4B (OH[−]) and evapd *in vacuo* to give the sugar moiety. The sugar was converted to its TMSi

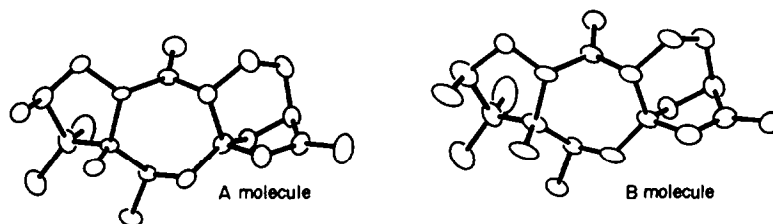


Fig. 1. Stereoscopic views of the structures of molecules A and B of **4**. The figures were drawn by an ORTEP program by Johnson [3].

derivative and identification was made by comparison of *R_f* of the authentic TMSi-D-glucose by GLC. GLC was performed at 170° using a stainless column (2m×3mm) of 5% OV-1 on Chromosorb W(AW).

Enzymatic hydrolysis of 1a. To a soln of 1a (100 mg) dissolved in HOAc–NaOAc buffer (pH 4.1, 10 ml) was added crude naringinase 'Sankyo' (100 mg) and incubated at 37°. After 24 hr, 50 mg of naringinase was added and further incubated for 24 hr. The soln was extracted with EtOAc and purified by Si gel PLC (eluent: CHCl₃–MeOH, 9:1) to give the aglycone (2a, 39 mg). Mp 232–233° (decomp.) (EtOAc). (Found: C, 71.37; H, 9.73. Calc. for C₂₀H₃₂O₄; C, 71.39; H, 9.59%). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3370, 1635. ¹H NMR (CDCl₃): δ 0.98, 1.24, 1.38 (each 3H, s), 3.58–3.78 (2H, m), 5.18 (2H, br s). ¹³C NMR (C₅D₅N): δ 21.1 (Me), 22.8 (Me), 24.8 (Me), 26.6 (C-11 or C-12), 26.8 (C-12 or C-11), 35.6 (C-14), 36.8 (C-2), 45.7 (C-8), 48.8 (C-7), 49.6 (—CH—), 50.8 (C-4), 55.7 (—CH—), 57.3 (C-15), 58.7 (—CH—), 71.1 (C-6), 78.6 (C-16), 83.6 (C-3), 86.8 (C-5), 110.5 (—C=CH₂), 153.9 (—C=CH₂).

Acetylation of 2a. Treatment of 2a (11 mg) with Ac₂O–Py overnight at room temp. gave a diacetate (2b) (12 mg), which was purified by Si gel PLC (eluent: CHCl₃–MeOH, 19:1). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3490, 1730, 1632, 1245. ¹H NMR (CDCl₃): δ 0.92, 1.08, 1.37 (each 3H, s), 2.06, 2.11 (each 3H, s), 4.84 (1H, d, *J* = 10 Hz), 5.02 (1H, dd, *J* = 6 and 10 Hz), 5.19 (1H, br s), 5.27 (1H, br s). MS *m/e*: 360 (M⁺–HOAc), 342 (M⁺–HOAc–H₂O), 300 (M⁺–2HOAc), 282 (M⁺–2HOAc–H₂O).

Dehydration of 2a. To a soln of 2a (52 mg) in dioxane (5 ml) was added dry CuSO₄ (50 mg) and refluxed for 8 hr. After filtration of CuSO₄, the filtrate was evapd *in vacuo* and purified by Si gel PLC (eluent: CHCl₃–MeOH, 9:1) to yield 4 (34 mg). Mp 205–206° (decomp.) (EtOAc). (Found: C, 75.60; H, 9.68. Calc. for C₂₀H₃₀O₃; C, 75.43; H, 9.50%). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3325, 1630. ¹H NMR (C₅D₅N): δ 1.12, 1.57 (each 3H, s), 1.62 (1H, d, *J* = 2 Hz), 3.00 (2H, m), 3.80 (1H, m), 3.93 (1H, m), 5.07 (1H, d, *J* = 2 Hz), 5.13, 5.23 (each 1H, s). MS *m/e*: 318 (M⁺), 300 (M⁺–H₂O), 282 (M⁺–2H₂O).

X-ray analysis. Crystals grown from EtOAc solns were colourless prisms elongated along the *c*-axis. The lattice constants and intensity data were collected on a Philips PW 1100 diffractometer using CuK α radiation monochromated by a graphite plate.

Crystal data. C₂₀H₃₀O₃, MW = 318.5, orthorhombic, space group P2₁2₁2₁, *Z* = 8, *D_x* = 1.189 g/cm³, *a* = 32.137 (15), *b* = 15.150 (7), *c* = 7.308 (4) Å, *U* = 3558.1 Å³. The unit cell contains two crystallographically independent molecules A and B. Intensities of 3832 independent structure

factors were obtained as above the 2 σ (I) level within the 2 θ angle of 156°. The crystal structure was solved by the direct method using MULTAN [9] and refined by the block-diagonal least-squares calculations using HBL5 [10]. The atomic coordinates for this work are available on request from the Director of the Cambridge Crystallographic Data Centre, University Chemical Laboratory, Lensfield Road, Cambridge, CB2 1EW, U.K.

Reaction of 2a with methanesulfonyl chloride. 2a (30 mg) was dissolved in Py (1 ml) and a small amount of MsCl (3 drops) was added. After 15 min at room temp., H₂O was added and extracted with EtOAc. The EtOAc layer was washed with H₂O, 5% KOH soln and H₂O, successively. After removal of the solvent, the residue was purified by Si gel PLC (eluent: CHCl₃–MeOH, 9:1) to yield 13 mg of colourless crystals (5). Mp 245° (EtOAc). (Found: C, 75.35; H, 9.76. Calc. for C₂₀H₃₀O₃; C, 75.43; H, 9.50%). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3495, 3385, 1702, 1640. ¹H NMR (C₅D₅N): δ 1.16, 1.46, 1.54 (each 3H, s), 4.18 (1H, t, *J* = 3 Hz), 4.89, 5.08 (each 1H, d, *J* = 1 Hz). MS *m/e*: 318 (M⁺), 300 (M⁺–H₂O), 282 (M⁺–2H₂O). CD [θ]_{290.5} = 12 500 (MeOH, *c* = 4.6 × 10⁻⁴).

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REFERENCES

1. Sakakibara, J., Shirai, N., Kaiya, T. and Nakata, H. (1978) *Phytochemistry* **17**, 1672.
2. Sakakibara, J., Shirai, N., Kaiya, T. and Nakata, H. (1979) *Phytochemistry* **18**, 135.
3. Johnson, C. K. (1965) *ORTEP Rep. ORNL-3794*. Oak Ridge National Laboratory, Tennessee.
4. Hikino, H., Koriyama, S. and Takemoto, T. (1973) *Tetrahedron* **29**, 773.
5. Furusaki, A., Hamanaka, N., Miyakoshi, H., Okuno, T. and Matsumoto, T. (1972) *Chem. Letters* 783.
6. Kasai, R., Suzuo, M., Asakawa, J. and Tanaka, O. (1977) *Tetrahedron Letters* 175.
7. Seo, S., Tomita, Y., Tori, K. and Yoshimura, Y. (1978) *J. Am. Chem. Soc.* **100**, 3331.
8. Furusaki, A., Gasa, S., Hamanaka, N., Ikeda, R. and Matsumoto, T. (1979) *Chem. Letters* 665.
9. Main, P., Woolfson, M. M. and Germain, G. (1971) *MULTAN. A Computer Program for the Automatic Solution of Crystal Structures*. Universities of York, England and Louvain, Belgium.
10. Okaya, Y. and Ashida, T. (1967) *HBL5 IV. The Universal Crystallographic Computing System (I)* p. 65. The Crystallographic Society of Japan, Tokyo.